



Monodentate *S*-vs. bidentate 1,5-*O,S*-coordination of *N*-phosphoryl-*N'*-(*R*)-thioureas with Pd(II)

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ABSTRACT

Reaction of *N*-phosphorylated thioureas of common formula $\text{RNHC(S)NHP(O)(OiPr)}_2$ ($\text{R} = t\text{Bu}$, **HL**^I; $\text{R} = \text{Ph}$, **HL**^{II}; $\text{R} = 4'$ -benzo-15-crown-5, **HL**^{III}) with $\text{Pd(PhCN)}_2\text{Cl}_2$ in acetonitrile leads to complexes of the structure $\text{Pd(HL}^{\text{I-III}}\text{-S)}_2\text{Cl}_2$ (**1**), $\text{Pd(HL}^{\text{II}}\text{-S)}_2\text{Cl}_2$ (**2**) and $\text{Pd(HL}^{\text{III}}\text{-S)}_2\text{Cl}_2$ (**3**). Reaction of *N*-phosphorylated thioureas of common formula $\text{RC(S)NHP(O)(OiPr)}_2$ ($\text{R} = \text{Et}_2\text{N}$, **HL**^{IV}; $\text{R} = \text{morpholine-}N\text{-yl}$, **HL**^V) in the same conditions leads to complexes $\text{Pd(L}^{\text{IV-V}}\text{-O,S)}_2$ (**4**) and $\text{Pd(L}^{\text{V}}\text{-O,S)}_2$ (**5**), where the palladium(II) atoms are coordinated in a square-planar fashion by the C=S sulfur atoms and the P=O oxygen atoms of two anionic ligands. The crystal structure of complex **1** has been investigated by X-ray crystallography. It was established that the thiourea ligands are in a *trans*-configuration and the palladium(II) cation is coordinated by the sulfur atoms of the C=S groups and the chlorine atoms. Complex **1** is the first example of palladium(II) complex in which the potentially chelating *N*-phosphorylated thiourea ligand is bound through the sulfur atom only.

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1. Introduction

The coordination chemistry of polyfunctional ligands, capable to realize different coordination modes with metal cations [1] is of interest for synthesis of new selective complexing agents and analytical reagents. Koch and others have shown that the coordination of *N,N*-dialkyl-*N'*-aroylthioureas leads to bidentate-*O,S*-coordination to Pd(II) [2,3]. It was also shown that the coordination mode of *N*-alkyl-*N'*-aroylthioureas depends on an intramolecular hydrogen bond between the donor oxygen atom of the carbonyl group and the hydrogen atom of the NH group [2].

We have become interested in the coordination chemistry of *N*-phosphorylated thioureas $\text{RR'NC(S)NHP(O)(OiPr)}_2$ ($\text{R} = \text{R}' = \text{alkyl}$; $\text{R} = \text{H}$, $\text{R}' = \text{alkyl, aryl}$) towards various transition metal cations [4] in view of their application in metals separation [5]. Recently, we reported, that in the square-planar complexes of *N*-phosphorylthioureas $\text{M[RNHC(S)NP(O)(OiPr)}_2\text{]}_2$, where ($\text{M} = \text{Ni(II)}$, $\text{R} = \text{Ph}$ [6], $p\text{-MeOC}_6\text{H}_4$, $p\text{-BrC}_6\text{H}_4$, $t\text{Bu}$, $i\text{Pr}$, $c\text{-Hex}$ [7]; $\text{M} = \text{Pd(II)}$, $\text{R} = \text{Ph}$ [6], $i\text{Pr}$ [7]; $\text{M} = \text{Cu(II)}$, $\text{R} = \text{Ph}$, $c\text{-Hex}$ [6]), 1,3-*N,S*-coordination of the anionic ligand is realized.

Recently, we reported some synthetic results presented here as a preliminary communication [8]. Herein, we report the comparative study on the complexes of the Pd(II) ions with *N*-phospho-

rylthioureas, containing secondary (Scheme 1) and tertiary (Scheme 2) nitrogen at the *N*-thioacylamidophosphate moiety $\text{C(S)NHP(O)(OiPr)}_2$. The crystal structure of *trans*- $\text{Pd(HL}^{\text{I}}\text{-S)}_2\text{Cl}_2$, the first example of *N*-phosphorylthiourea coordinated to Pd(II) only through the S atom in the monodentate fashion, is also reported.

2. Experimental

2.1. Synthesis of **HL**^I–**HL**^V

N-Diisopropoxyphosphorylthioureas were prepared according to the previously described methods [9].

2.2. Synthesis of **1–5**

A portion of $\text{Pd(PhCN)}_2\text{Cl}_2$ (0.192 g, 0.5 mmol) dissolved in 30 mL of acetonitrile was added dropwise to 30 mL of a well stirred solution containing **HL**^{I–V} (0.296, 0.316, 0.506, 0.295 or 0.310 g; 1 mmol) in the same solvent. The mixture was stirred at room temperature for a further 10 h. The solvent was then removed *in vacuo*. A precipitate was isolated from dichloromethane by *n*-hexane.

2.2.1. $\text{Pd(HL}^{\text{I}}\text{-S)}_2\text{Cl}_2$ (**1**)

Yield: 0.331 g (86%). M.p. 128–129 °C. ¹H NMR (CDCl_3): δ 1.38–1.43 (m, 12H, CH_3), 1.49–1.53 (m, 9H, CH_3 , $t\text{Bu}$), 4.58–4.73 (m, 2H,

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